

## (R)-4-MENTHEN-3-ONE *anti*-OXIME AND ITS TRANSFORMATION UNDER BECKMAN REARRANGEMENT CONDITIONS

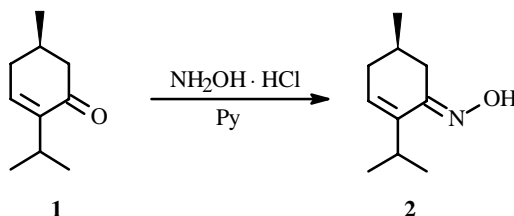
R. Ya. Kharisov,<sup>1</sup> E. R. Latypova,<sup>2</sup> R. F. Talipov,<sup>2</sup>  
R. R. Muslukhov,<sup>1</sup> G. Yu. Ishmuratov,<sup>1</sup>  
and G. A. Tolstikov<sup>1</sup>

UDC 542.952.3+547.388.4+547.596.4

*(R)*-4-Menthenone *anti*-oxime was synthesized for the first time. Its transformations under Beckman rearrangement conditions were studied.

**Key words:** *anti*-oxime, Beckman rearrangement, (*R*)-4-menthenone, (*S*)-3,7-dimethyl-6-oxooctanoic acid methyl ester, Semmler—Wolf reaction.

The synthesis of a totally optically pure chiral unit that was based on ozonolytic decyclization of (*R*)-4-menthenone (**1**) was reported previously [1, 2]. In order to expand the use of **1**, we prepared its oxime **2** (Scheme 1), which was synthesized previously [3] by the reaction of racemic menthene and nitrosyl chloride with subsequent dehydrochlorination. It was proposed to have the *syn*-configuration. However, the stereochemistry of the ketoxime could not be unambiguously determined using the traditional Beckman rearrangement. The *syn*-structure of the oxime was confirmed only later [4].

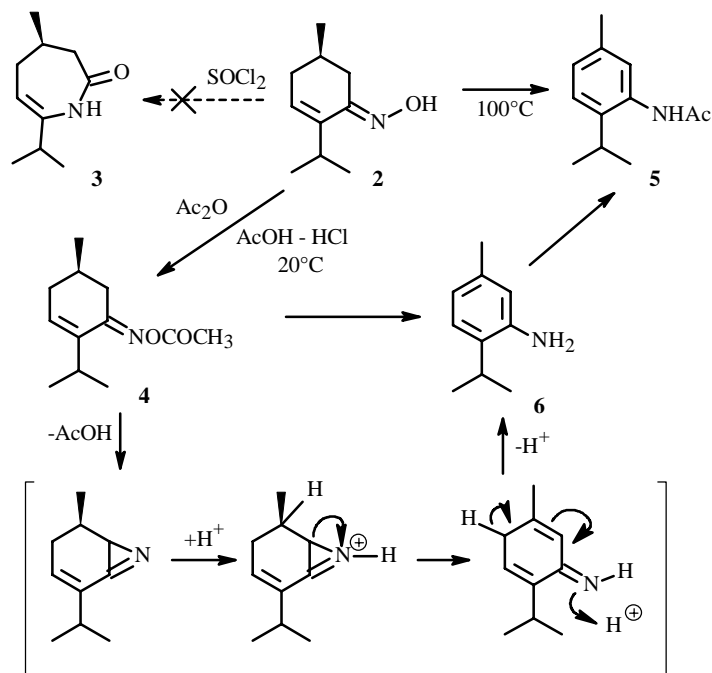


Scheme 1.

A comparison of the mp (57–58°C) and UV spectrum (EtOH,  $\lambda_{\text{max}}$  232 nm,  $\log \epsilon$  3.55) of **2** synthesized by us with data for the compound prepared earlier (mp 66–67°C,  $\lambda_{\text{max}}$  242 nm,  $\log \epsilon$  4.1 [4]) showed that they differ substantially.

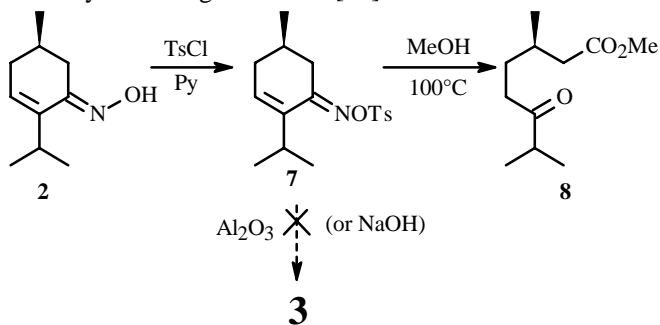
Based on these differences and the fact that *syn*-oximes usually have higher melting points [5], we assigned the *anti*-configuration of the hydroxyl and double bond to our **2**. Furthermore, it is known that *anti*-oximes, with rare exceptions, either decompose to tars or do not react under Beckman-rearrangement conditions [6]. Oxime **2** would be converted to tetrahydroazepinone **3** through a successful reaction. However, it did not rearrange upon treatment with thionyl chloride and decomposed upon treatment with  $\text{P}_2\text{O}_5$  and conc.  $\text{H}_2\text{SO}_4$ . Use of the Beckman mixture ( $\text{Ac}_2\text{O}$ — $\text{AcOH}$ — $\text{HCl}$ ) at various temperatures (20 and 100°C) produced the O-acyl derivative **4** and the product of its further aromatization, acetamide **5**. This can be explained by the initial formation of **4**, which then was transformed further (according to Semmler—Wolf) into amine **6** with subsequent acylation to **5** (Scheme 2).

1) Institute of Organic Chemistry, Ufa Scientific Center, Russian Academy of Sciences, 450054, Ufa, pr. Oktyabrya, 71, fax: (3472) 35 60 66, e-mail: kharis@anrb.ru; 2) Bashkir State University, 450074, Ufa, ul. Frunze, 32, fax: (3472) 22 61 05, e-mail: TalipovRF@bsu.bashedu.ru. Translated from *Khimiya Prirodnikh Soedinenii*, No. 6, pp. 470–472, November–December, 2003. Original article submitted November 3, 2003.



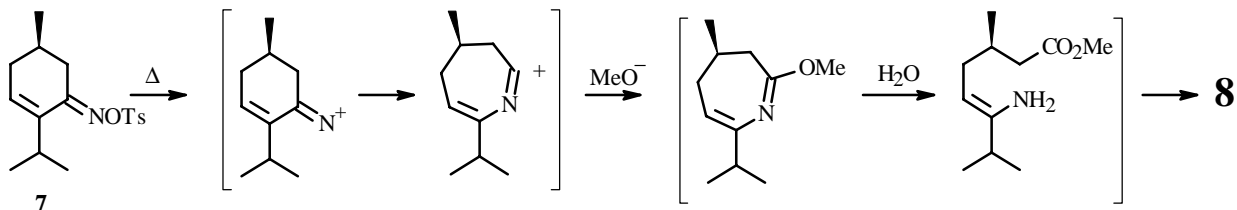
Scheme 2.

The majority of oxime arylsulfonates are exceedingly reactive compounds that undergo the Beckman rearrangement during synthesis [7]. However, reaction of **2** with *p*-toluenesulfonyl chloride produced O-tosyl derivative **7**, which was unreactive toward several traditional reagents, for example, aluminum oxide [8] and  $\text{NaOH}$  (in  $\text{THF-H}_2\text{O}$ ) [9]. Also, heating ( $100^\circ\text{C}$ ) in  $\text{MeOH}$  formed a complicated mixture from which we isolated using column chromatography (*S*)-3,7-dimethyl-6-oxooctanoic acid methyl ester (**8**) in 70% yield (Scheme 3). Compound **8** was identical to that prepared previously by us from menthone through an intermediate Baeyer—Villiger reaction [10].



Scheme 3.

The proposed reaction mechanism is consistent with already known transformations of oxime tosylates [7] (Scheme 4).



Scheme 4.

Thus, we prepared for the first time the *anti*-oxime of optically pure menthenone. For this, the Beckman rearrangement could be used successfully only for the corresponding tosylate. The reaction occurs with deazotization, giving ketoester **8**.

## EXPERIMENTAL

IR spectra were recorded on a Specord M-82 instrument as thin layers. NMR spectra ( $\delta$ , ppm, J/Hz) were obtained on a Bruker AM-300 spectrometer (working frequency 300.13 MHz for PMR and 75.47 MHz for  $^{13}\text{C}$ ) in  $\text{CDCl}_3$  relative to TMS. Signals in PMR spectra were assigned and spin—spin coupling constants (SSCC) were determined using double resonance and two-dimensional correlation spectroscopy (COSY-H-H). Mass spectra were measured in an MX-1320 instrument at ionizing potential 70 eV. UV spectra were recorded on a Specord M400 instrument. Chromatographic analysis was performed on a Chrom-5 instrument [column length, 2.4 m; stationary phase, PEG-6000 (5%) on Inerton AW-DMCS (0.125-0.160 mm); working temperature 50–200°C] with He carrier gas. Optical rotations were measured on a Perkin—Elmer-241-MC polarimeter. Solvents were dried as usual. Column chromatography was performed over  $\text{SiO}_2$  (L, 60–200  $\mu\text{m}$ , Lancaster, England). TLC monitoring was carried out over  $\text{SiO}_2$  (Silufol, Czech Rep.). Petroleum ether (PE) (bp 40–70°C) was used for chromatography.

Elemental analyses of all compounds agreed with those calculated.

**(R)-5-Methyl-2-(1-methylethyl)-2-cyclohexen-1-one Oxime (2).** A solution of **1** (2.00 g, 13.2 mmol) in Py (22 mL) was stirred (20°C) and treated with  $\text{NH}_2\text{OH}\cdot\text{HCl}$  (4.40 g, 55.7 mmol). After 1 h the reaction mixture was diluted with ethylacetate (50 mL), washed with  $\text{H}_2\text{O}$  ( $2 \times 50$  mL), dried over  $\text{Na}_2\text{SO}_4$ , and evaporated. The solid was recrystallized from aqueous EtOH (50%) to afford **2** (1.66 g, 77%) as white crystals,  $\text{C}_{10}\text{H}_{17}\text{NO}$ , mp 57–58°C,  $[\alpha]_{\text{D}}^{19} -47.9^\circ$  ( $c$  9.99,  $\text{CHCl}_3$ ),  $R_f$  0.6 (PE:MeO-*t*-Bu, 2:1).

UV spectrum (EtOH,  $\lambda_{\text{max}}$ , nm): 232 (log  $\epsilon$  3.55) [4].

IR spectrum (KBr,  $\nu$ ,  $\text{cm}^{-1}$ ): 964 (N–O), 1636 (C=N), 3268 (O–H).

PMR ( $\delta$ , ppm, J/Hz): 1.01 (d, 3H,  $J = 5.5$ ,  $\text{CH}_3\text{C}-5$ ), 1.07 [d, 6H,  $J = 6.9$ ,  $(\text{CH}_3)_2\text{C}$ ], 1.70–1.90 (m, 2H,  $\text{H}_a-4$ , H-5), 1.92 (d, 1H,  $J = 13.3$ ,  $\text{H}_a-6$ ), 2.23 (dd, 1H,  $^2J = 13.8$ ;  $^3J = 9.0$ ,  $\text{H}_e-4$ ), 2.80 (sept., 1H,  $J = 6.9$ , HCC-2), 3.12 (d, 1H,  $^2J = 13.3$ ,  $\text{H}_e-6$ ), 6.01 (d, 1H,  $J = 4.5$ , H-3), 9.30 (br.s, 1H, O–H).

$^{13}\text{C}$  NMR ( $\delta$ , ppm): 21.27 ( $\underline{\text{C}}\text{H}_3\text{C}-5$ ), 21.29 and 22.39 [ $\underline{\text{C}}(\text{H}_3)_2\text{C}$ ], 27.03 (C-8), 27.78 (C-5), 30.81 (C-4), 33.42 (C-6), 129.19 (C-3), 140.14 (C-2), 155.51 (C-1).

**(R)-5-Methyl-2-(1-methylethyl)-2-cyclohexen-1-one Tosylate Oxime (7).** A mixture of **2** (0.35 g, 2.1 mmol) and dry Py (10 mL) at 0°C was treated with *p*-TsCl (0.44 g, 2.3 mmol), stirred and heated to room temperature, and left for 18 h. The Py was evaporated in vacuum. The solid was dissolved in  $\text{H}_2\text{O}$  (10 mL) and extracted with  $\text{CH}_2\text{Cl}_2$  ( $3 \times 30$  mL). The extract was washed successively with  $\text{H}_2\text{O}$  and saturated NaCl and  $\text{NaHCO}_3$  solutions, dried over  $\text{MgSO}_4$ , and evaporated to afford crude product (0.56 g), recrystallization of which from aqueous EtOH (50%) isolated **7** (0.51 g, 76%) as white crystals, mp 95.5–96.5°C,  $R_f$  0.63 (PE:MeO-*t*-Bu, 2:1),  $[\alpha]_{\text{D}}^{18} -20.4^\circ$  ( $c$  1.57,  $\text{CHCl}_3$ ).

IR spectrum (KBr,  $\nu$ ,  $\text{cm}^{-1}$ ): 952 (N–O); 1192, 1366 (S=O); 1492, 1594 (C=C); 1636 (C–N).

PMR ( $\delta$ , ppm, J/Hz): 0.96 (d, 3H,  $^3J = 5.5$ ,  $\text{H}_3\text{CC}-5$ ), 0.99 [d, 6H,  $^3J = 5.9$ ,  $(\text{CH}_3)_2\text{C}$ ], 1.70–1.83 (m, 2H,  $\text{H}_a-4$ , H-5), 1.94 (d, 1H,  $^2J = 12.9$ ,  $\text{H}_a-6$ ), 2.25 (dd, 1H,  $^2J = 15.3$ ,  $^3J = 7.2$ ,  $\text{H}_e-4$ ), 2.44 (s, 3H,  $\text{H}_3\text{C}-\text{Ar}$ ), 2.76 (g, 1H,  $^3J = 5.9$ , HCC-2), 6.19 (dd, 1H,  $J = 6.2$ ,  $J = 2.3$ , H-3), 7.33 and 7.89 (both d, 4H,  $^3J = 8.3$ , H-Ar).

### Beckman Rearrangement Methods. Preparation of **4**, **5**, and **8**.

**A.** Dry HCl was bubbled through a solution of **2** (0.64 g, 3.8 mmol) in a mixture of glacial AcOH (3.7 mL) and  $\text{Ac}_2\text{O}$  (1.85 mL) at 20°C for 1 h. The reaction mixture was left for 43 h, treated with  $\text{H}_2\text{O}$  (10 mL), extracted with  $\text{CH}_2\text{Cl}_2$  ( $3 \times 20$  mL), dried over  $\text{MgSO}_4$ , and evaporated. The solid (0.70 g) was chromatographed over  $\text{SiO}_2$  (eluent PE:MeO-*t*-Bu, 10:1) to afford **4** (0.45 g, 64%).

**B.** Dry HCl was bubbled until saturated (45 min) through a solution of **2** (1.00 g, 5.99 mmol) in a mixture of glacial AcOH (9.52 mL, 9.99 g, 1.667 mmol) and  $\text{Ac}_2\text{O}$  (1.85 mL, 19.6 mmol) in a glass ampul. The ampul was sealed and heated for 3 h on a boiling-water bath. The reaction mixture was left for 48 h at room temperature, diluted with  $\text{H}_2\text{O}$  (5 mL), and extracted with  $\text{Et}_2\text{O}$  ( $3 \times 30$  mL). The extract was washed with saturated  $\text{NaHCO}_3$  solution (until the pH was 7), dried over  $\text{MgSO}_4$ , and evaporated. The solid (0.94 g) was chromatographed over  $\text{SiO}_2$  (eluent PE:MeO-*t*-Bu, 10:1) to afford **5** (0.25 g, 27%).

C. A solution of **7** (2.53 g, 7.88 mmol) in absolute MeOH (50 mL) was heated in an ampul for 4.5 h on a boiling-water bath, evaporated in vacuum, treated with aqueous NaOH solution (10%, until the pH was ~10), extracted with CH<sub>2</sub>Cl<sub>2</sub>, dried over Na<sub>2</sub>SO<sub>4</sub>, and evaporated. The solid (1.58 g) was chromatographed (eluent PE:CH<sub>2</sub>Cl<sub>2</sub>, 5:1) to afford **8** (1.10 g, 70%) [10].

**(R)-6-[(Acetyloxy)imino]-4-methyl-1-isopropylcyclohex-1-ene (4)**. *R<sub>f</sub>* 0.56 (PE:MeO-*t*-Bu, 2:1), [ $\alpha$ ]<sub>D</sub><sup>24</sup> -52.8° (*c* 3.08, CHCl<sub>3</sub>), C<sub>12</sub>H<sub>19</sub>NO<sub>2</sub>.

IR spectrum (KBr,  $\nu$ , cm<sup>-1</sup>): 1640 (C=C), 1685 (C=N), 1775 (C=O).

PMR ( $\delta$ , ppm, J/Hz): 0.93 (d, 3H, <sup>3</sup>J = 5.4, CH<sub>3</sub>C-5), 1.03 [d, 6H, J = 6.9, (CH<sub>3</sub>)<sub>2</sub>C], 1.70-1.85 (m, 2H, H<sub>a</sub>-4, H-5), 1.90 (d, 1H, <sup>2</sup>J = 11.8, H<sub>a</sub>-6), 2.23 (dd, 1H, <sup>2</sup>J = 12.8, <sup>3</sup>J = 6.3, H<sub>c</sub>-4), 2.18 (s, 3H, CH<sub>3</sub>CO), 3.03 (d, 1H, <sup>2</sup>J = 11.8, H<sub>e</sub>-6), 6.18 (d, 1H, J = 4.3, H-3).

<sup>13</sup>C NMR ( $\delta$ , ppm): 19.81 (CH<sub>3</sub>CO), 20.94 and 21.81 [(CH<sub>3</sub>)<sub>2</sub>C], 22.18 (CH<sub>3</sub>C-5), 26.97 (HCC-2), 27.67 (C-5), 32.24 (C-4), 33.24 (C-6), 133.15 (C-3), 139.97 (C-2), 161.31 (C-1), 169.60 (C=O).

Mass spectrum (EI, 70 eV, *m/z*, *I*<sub>rel</sub>, %): 209 (0.8) [M]<sup>+</sup>, 167 (3) [M - CH<sub>2</sub>CO]<sup>+</sup>, 150 (32) [M - CH<sub>3</sub>COO]<sup>+</sup>, 148 (19), 134 (54), 107 (54), 94 (21), 93 (23), 91 (14), 81 (17), 79 (21), 77 (18), 67 (31), 65 (11), 55 (20), 53 (20), 43 (100) [CH<sub>3</sub>CO]<sup>+</sup>, 42 (22), 41 (69), 39 (35), 27 (33).

**5-Methyl-2-isopropylacetanilide (5)**. *R<sub>f</sub>* 0.36 (CH<sub>2</sub>Cl<sub>2</sub>).

IR spectrum (KBr,  $\nu$ , cm<sup>-1</sup>): 1490, 1600 (C=C); 1555, 3310 (N-H); 1685 (C=O).

PMR ( $\delta$ , ppm, J/Hz): 1.20 [d, 6H, J = 6.72, (CH<sub>3</sub>)<sub>2</sub>C], 2.17 (s, 3H, CH<sub>3</sub>CO), 2.30 (s, 3H, CH<sub>3</sub>C-5), 2.96-3.08 (m, 1H, HC), 6.93-7.21 (m, H-Ar).

<sup>13</sup>C NMR ( $\delta$ , ppm): 19.55 (CH<sub>3</sub>CO), 20.83 (CH<sub>3</sub>C-5), 23.15 and 23.91 [(CH<sub>3</sub>)<sub>2</sub>C], 27.58 (HC), 125.41 (C-3), 126.11 (C-6), 127.18 (C-4), 133.61 (C-1), 135.85 (C-5), 138.41 (C-2), 169.05 (C=O).

**(S)-3,7-Dimethyl-6-oxooctanoic acid methyl ester (8)**. *R<sub>f</sub>* 0.5 (PE:MeO-*t*-Bu, 2:1), [ $\alpha$ ]<sub>D</sub><sup>18</sup> +9.5° (*c* 0.16, CHCl<sub>3</sub>) [10].

The IR spectrum is practically identical to that described previously [10].

PMR ( $\delta$ , ppm, J/Hz): 0.95 (d, 3H, <sup>3</sup>J = 6.6, CH<sub>3</sub>C-3), 1.09 [d, 6H, <sup>3</sup>J = 6.9, (CH<sub>3</sub>)<sub>2</sub>C], 1.40-1.54 (m, 1H, H-4), 1.56-1.68 (m, H, H'-4), 1.87-2.00 (m, 1H, H-3), 2.13 (dd, 1H, <sup>2</sup>J = 14.8, <sup>3</sup>J = 8.1, H'-2), 2.31 (dd, 1H, <sup>2</sup>J = 14.8, <sup>3</sup>J = 5.9, H-2), 2.43-2.50 (m, 2H, H-5), 2.61 (g, 1H, H-7), 3.68 (s, 3H, CH<sub>3</sub>O).

<sup>13</sup>C NMR ( $\delta$ , ppm): 18.26 [(CH<sub>3</sub>)<sub>2</sub>C-6], 19.51 (CH<sub>3</sub>C-3), 30.01 (C-3), 30.29 (C-4), 37.81 (C-5), 40.84 (C-7), 41.39 (C-2), 51.45 (CH<sub>3</sub>O), 173.36 (C-1), 214.49 (C-6).

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